Tetrahedron Letters No. 34, pp 2919 - 2922, 1977. Pergamon Press. Printed in Great Britain.

2-CARBOMETHOXY-4,4-DIMETHYL-2,5-CYCLOHEXADIEN-1-ONE AS A DIENOPHILE. A CONVENIENT APPROACH TO THE 4,4-DIMETHYL-1-DECALONE SYSTEM

H. J. Liu^{*} and E. N. C. Browne Department of Chemistry, University of Alberta Edmonton, Alberta, Canada, T6G 2G2

(Received in USA 5 May 1977; received in UK for publication 27 June 1977)

In approaches to the total synthesis of polycyclic di- and triterpenoids¹, the cycloaddition of 4,4-dimethyl-2-cyclohexen-l-one $(\underline{1})^2$ to a suitably functionized vinylcyclohexene $\underline{2}$ represents an especially attractive general approach. As shown in Scheme 1, such a process not only facilitates the formation of the <u>gem-dimethyl containing tricycle 3</u> (common to a vast number of higher terpenoids¹) with suitable functionalities for further elaboration, it also predetermines the relative stereochemistry of two critical chiral centers C-5 and C-9³ in the desired orientation. Obviously, the feasibility of this approach relies on the reactivity of $\underline{1}$ as a dienophile. In this connection, the Diels-Alder reactions of $\underline{1}$ with isoprene and 1,3-butadiene were examined under a variety of conditions. In no instance however, did the attempted reaction give rise to more than a trace (<5% yield) of the desirable adduct. As a consequence of these findings, several derivatives of $\underline{1}$ were subjected to further investigation.



Scheme 1

In sharp contrast to the lack of dienophilicity observed for its parent compound $\underline{1}$, the titled dienone ester $\underline{4}^4$ was found to be a powerful dienophile⁵. In the presence of boron trifluoride, the cycloaddition of $\underline{4}$ to the four dienes studied (see Table 1) proceeded smoothly and cleanly under mild reaction conditions similar to those given below for cyclopentadiene.

A solution of $\underline{4}$ (297 mg; 1.65 mmol) and boron trifluoride etherate (116 mg; 0.85 mmol) in ether (20 ml) was stirred at room temperature for 20 min. Cyclopentadiene (2.17 g; 33 mmol)⁶ was added as a solution in ether (5 ml). The resulting solution, after stirring for 3 hr, was made basic with aqueous sodium bicarbonate and extracted with methylene chloride. The organic solution, after work-up in the usual manner gave an oil which was chromatographed on silica gel. Elution with a solution of 5% ether in hexane afforded 284 mg of $\underline{5}$: ir (neat) 1738, 1662, 1379 and 1366 cm⁻¹; nmr (CCl₄) δ 1.17, 1.27 (both s, 3H each, $-\dot{C}$ (CH₃)₂), 1.41 (ddd, J = 9, J' = J" = 2 Hz, 1H, -CHH-), 1.60 (d, J = 9 Hz, 1H, -CHH-), 2.49 (dd, J = 2.5, J' = 1.5 Hz, 1H, (CH₃)₂ \dot{C} CH⁻), 3.00, 3.45 (both m, 1H each, $-\dot{C}$ HCH₂ \dot{C} H⁻), 3.65 (s, 3H, -OCH₃), 5.60 (d, J = 10 Hz, 1H, -COCH=), 5.75 6.05 (both dd, J = 6, J' = 2.5 Hz, 1H each, -CH=CH-), and 6.31 (dd, J = 10, J' = 1.5 Hz, 1H, -COCH=CH-); ms M⁺ 246.1253 (calcd. for C₁₅H₁₈O₃ = 246.1256).

Diene	Time (hr)	Product ⁷	Isolated % yield
	3	COOCH ₃	70
	6	COOCH ₃	70
1	4	COOCH3	90
X	96	COOCH3	63

Table 1. Cycloaddition of 4 to 1,3-dienes

No. 34

While <u>4</u> possesses two double bonds, both theoretically subject to diene attack, the cycloaddition reactions were shown to be completely regioselective. Without exception, the compounds listed in Table 1 were obtained as the only adducts. The exclusive participation of the more substituted double bond is apparently due to enhancement of its reactivity by the electron-withdrawing ester group.⁹

The foregoing report illustrates the first examples of the direct formation of synthetically very useful 4,4-dimethyl-1-decalone derivatives using a Diels-Alder approach. This method for the construction of the gem-dimethyl substituted decalin system compares favorably to those existing¹¹ in terms of both simplicity and efficiency. It is also noteworthy that this method permits direct functionalization of the aforementioned system at C-1, a salient feature which would otherwise be difficult to achieve. Its extension to terpenoid synthesis is in progress.

Acknowledgement: We thank the National Research Council of Canada and the University of Alberta for financial support.

References and Footnotes

- T. K. Devon and A. I. Scott, "Handbook of Naturally Occurring Compounds," Vol. II, Academic Press, New York, N.Y., 1972.
- R. L. N. Harris, F. Komitsky, Jr., and C. Djerassi, J. Amer. Chem. Soc., 89, 4765 (1967).
- Steroidal numbering is used throughout for all the compounds possessing two or more rings.
- Prepared in 53% yield from 1 in two steps: Carbomethoxylation [NaH and CO(OCH₃)₂ in refluxing 1,2-dimethoxyethane] and oxidation [SeO₂ in refluxing t-BuOH-HOAC (19:1)].
- 5. It was also found that 4,4-dimethyl-2,5-cyclohexadien-l-one underwent cycloadditions with dienes but at a much slower rate. Its addition to 1,3butadiene, for example, gave 18% yield of the desirable adduct with 32% of the starting material recovered after three weeks when the reaction was carried out under the conditions similar to those cited for <u>4</u> and cyclopentadiene (vide infra).
- In the case of 1,3-butadiene, the ether solution was saturated with the diene throughout the reaction.
- 7. The stereochemistry follows from the <u>cis</u> and <u>endo</u> rules⁸ governing the Diels-Alder reaction and is confirmed by nmr data.
- A. S. Onishchenko, "Diene Synthesis," Daniel Davy and Co., New York, N.Y., 1964.

- 9. A similar observation has been made previously on quinones possessing a strongly electron-withdrawing substituent¹⁰.
- 10. M. F. Ansell, G. C. Culling, B. W. Nash, D. A. Wilson, and J. W. Lown, Proc. Chem. Soc., 405 (1960).
- 11. See ref. 12-15 for examples of some frequently used methods.
- 12. H. O. House, "Modern Synthetic Reactions," 2nd ed., W. A. Benjamin, Inc., Menlo Park, Calif., 1972.
- G. Stork, P. Rosen, N. Goldman, R. V. Coombs, J. Tsuji, J. Amer. Chem. Soc., 87, 275 (1965).
- 14. W. S. Johnson, Accounts Chem. Res., 1, 1 (1968).
- 15. E. E. van Tamelen, Accounts Chem. Res., 1, 111 (1968).