2-CARBOMETHOXY-4,4-DIMETHYL-2,5-CYCLOHEXADIEN-1-ONE AS A DIENOPHILE. A CONVENIENT APPROACH TO THE<br>4,4-DIMETHYL-1-DECALONE SYSTEM<br>H. J. Liu* and E. N. C. Browne Department of Chemistry, University of Alberta Edmonton, Alberta, Canada, T6G 2G2

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In approaches to the total synthesis of polycyclic di- and triterpenoids ${ }^{1}$, the cycloaddition of 4,4-dimethyl-2-cyclohexen-1-one (1) ${ }^{2}$ to a suitably functionized vinylcyclohexene $\underline{2}$ represents an especially attractive general approach. As shown in Scheme l, such a process not only facilitates the formation of the gem-dimethyl containing tricycle $\underline{\underline{3}}$ (common to a vast number of higher terpenoids ${ }^{1}$, with suitable functionalities for further elaboration, it also predetermines the relative stereochemistry of two critical chiral centers C-5 and $C-9^{3}$ in the desired orientation. Obviously, the feasibility of this approach relies on the reactivity of $\underline{\underline{1}}$ as a dienophile. In this connection, the DielsAlder reactions of $\underline{\underline{l}}$ with isoprene and l,3-butadiene were examined under a variety of conditions. In no instance however, did the attempted reaction give rise to more than a trace ( $<5 \%$ yield) of the desirable adduct. As a consequence of these findings, several derivatives of $\underset{\sim}{1}$ were subjected to further investigation.


Scheme 1

In sharp contrast to the lack of dienophilicity observed for its parent compound 1 , the titled dienone ester $\underline{\underline{4}}^{4}$ was found to be a powerful dienophile ${ }^{5}$. In the presence of boron trifluoride, the cycloaddition of $\underline{\underline{4} \text { to the four dienes }}$ studied (see Table 1) proceeded smoothly and cleanly under mild reaction conditions similar to those given below for cyclopentadiene.

A solution of 4 ( $297 \mathrm{mg} ; 1.65 \mathrm{mmol}$ ) and boron trifluoride etherate (ll6 mg; 0.85 mmol ) in ether ( 20 ml ) was stirred at room temperature for 20 min . Cyclopentadiene ( $2.17 \mathrm{~g} ; 33 \mathrm{mmol})^{6}$ was added as a solution in ether ( 5 ml ). The resulting solution, after stirring for 3 hr , was made basic with aqueous sodium bicarbonate and extracted with methylene chloride. The organic solution, after work-up in the usual manner gave an oil which was chromatographed on silica gel. Elution with a solution of 5\% ether in hexane afforded 284 mg of 5 : ir (neat) 1738, 1662, 1379 and $1366 \mathrm{~cm}^{-1}$; $\mathrm{nmr}\left(\mathrm{CCl}_{4}\right) \delta 1.17,1.27$ (both $\mathrm{s}, 3 \mathrm{H}$ each, $\left.-{ }_{-}^{\prime}\left(\mathrm{CH}_{3}\right)_{2}\right), 1.41$ (ddd, $\left.J=9, J^{\prime}=J^{\prime \prime}=2 \mathrm{~Hz}, 1 \mathrm{H},-\mathrm{CHH}-\right), 1.60(\mathrm{~d}, \mathrm{~J}=9 \mathrm{~Hz}, 1 \mathrm{H}$,
 1 H each, $\left.-\stackrel{1}{\mathrm{C}} \mathrm{HCH}_{2} \stackrel{1}{\mathrm{C}}-\mathrm{H}-\right), 3.65\left(\mathrm{~s}, 3 \mathrm{H},-\mathrm{OCH}_{3}\right), 5.60(\mathrm{~d}, \mathrm{~J}=10 \mathrm{~Hz}, 1 \mathrm{H},-\mathrm{COCH}=), 5.75$ 6.05 (both dd, $J=6, J^{\prime}=2.5 \mathrm{~Hz}, 1 \mathrm{H}$ each, $-\mathrm{CH}=\mathrm{CH}-$ ), and $6.31(\mathrm{dd}, \mathrm{J}=10$, $J^{\prime}=1.5 \mathrm{~Hz}, 1 \mathrm{H},-\mathrm{COCH}=\mathrm{CH}-$ ) $; \mathrm{ms} \mathrm{M}^{+} 246.1253$ (calcd. for $\mathrm{C}_{15} \mathrm{H}_{18} \mathrm{O}_{3}=246.1256$ ).

Table 1. Cycloaddition of $\underline{\underline{4}}$ to 1,3 -dienes
Diene Time (hr)

While $\underline{\underline{4}}$ possesses two double bonds, both theoretically subject to diene attack, the cycloaddition reactions were shown to be completely regioselective. Without exception, the compounds listed in Table 1 were obtained as the only adducts. The exclusive participation of the more substituted double bond is apparently due to enhancement of its reactivity by the electron-withdrawing ester group. ${ }^{9}$

The foregoing report illustrates the first examples of the direct formation of synthetically very useful 4,4-dimethyl-l-decalone derivatives using a Diels-Alder approach. This method for the construction of the gem-dimethyl substituted decalin system compares favorably to those existing ${ }^{11}$ in terms of both simplicity and efficiency. It is also noteworthy that this method permits direct functionalization of the aforementioned system at $\mathrm{C}-1$, a salient feature which would otherwise be difficult to achieve. Its extension to terpenoid synthesis is in progress.

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## References and Footnotes

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3. Steroidal numbering is used throughout for all the compounds possessing two or more rings.
4. Prepared in $53 \%$ yield from 1 in two steps: Carbomethoxylation [NaH and $\mathrm{CO}\left(\mathrm{OCH}_{3}\right)_{2}$ in refluxing 1,2-dimethoxyethane] and oxidation [ $\mathrm{SeO}_{2}$ in refluxing t-BuOH-HoAc (19:1)].
5. It was also found that 4,4-dimethyl-2,5-cyclohexadien-l-one underwent cycloadditions with dienes but at a much slower rate. Its addition to 1,3butadiene, for example, gave $18 \%$ yield of the desirable adduct with $32 \%$ of the starting material recovered after three weeks when the reaction was carried out under the conditions similar to those cited for $\underline{\underline{4}}$ and cyclopentadiene (vide infra).
6. In the case of 1,3-butadiene, the ether solution was saturated with the diene throughout the reaction.
7. The stereochemistry follows from the cis and endo rules ${ }^{8}$ governing the Diels-Alder reaction and is confirmed by nmr data.
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